

Hingegen besteht eine Korrelation zwischen der acetylierenden Wirkung⁵ und dem im Blut enthaltenen Coenzym A. Aus einer Gruppe von 28 Untersuchungen erhält man einen Korrelationskoeffizienten von $r = +0,6244$ ($P \sim 0,001$). Diese Untersuchungen lassen uns zum Schluss kommen, dass das im Blut vorhandene Coenzym A zum grössten Teil in den Leukozyten und vielleicht auch in den Thrombozyten enthalten ist; hingegen ist der Gehalt im Plasma und in den Erythrozyten niedrig.

Diese Befunde stehen vielleicht in Beziehung zu kürzlich veröffentlichten Untersuchungen, wonach die Synthese der Lipide – ein Vorgang, in welchem das Coenzym A eine erste Rolle spielt – im zirkulierenden Blut vorwiegend in den Leukozyten und in geringerem Grade in den Thrombozyten stattfindet, während sie in den reifen Erythrozyten und im Plasma fast ganz fehlt^{6,7}.

Riassunto. Il Coenzima A nel sangue di soggetti leucemici, in particolare con leucemia mieloide, appare nettamente aumentato in confronto al sangue normale.

Il Coenzima A è contenuto in prevalenza nei leucociti e probabilmente anche nelle piastrine, mentre è scarso nel plasma e nei globuli rossi.

U. AMBANELLI, F. BONATI und G. SALVI

Istituto di Clinica Medica della Università di Parma und Laboratori Ricerche della Bracco Ind. Chimica, Milano (Italien), 18. Mai 1961.

⁵ μ g von acetyliertem PABA während 24 h in 1 ml Blut bei 37°.

⁶ P. MARKS, A. GELLHORN und C. KIDSON, *J. biol. Chem.* **235**, 2579 (1960).

⁷ E. ROWE, A. ALLISON und J. LOVELOCK, *Biochim. biophys. Acta* **41**, 310 (1960).

C16-Methyl Derivatives of Progesterone and 17 α -Acetoxypregesterone: Preparation and Biological Activities

In view of the recent publication¹ describing the preparation of certain C16-methyl steroids, we wish to report here the progestational activities associated with several derivatives of 16 α -methylprogesterone prepared in these Laboratories.

In Table I are given the comparative biological activities of 16 α -methylprogesterone² (I) and its 2 α -methyl (II)-, 6 α -methyl (III)³- and 6 α -fluoro (IV)-derivatives. Also included are the physical properties of the newly prepared compounds II and IV. 2 α , 16 α -Dimethylprogesterone (II) was obtained by alkylation⁴ of the C2-glyoxylic ester of 16 α -methylprogesterone (I). 6 α -Fluoro-16 α -methylprogesterone was prepared by employing either the epoxide-hydrofluoric acid⁵ or the N-bromoacetamide-hydrofluoric acid⁶ techniques for the introduction of the C6-fluorine

atom, each utilizing 16 α -methylpregnenolone² as starting material.

Summarized in Table II are the biological activities and physical properties of the 16 α -methyl derivatives of 17 α -hydroxypregesterone. 16 α -Methyl-17 α -hydroxypregester-

¹ E. BATRES, T. CARDENAS, J. A. EDWARDS, G. MONROY, O. MANCERA, C. DJERASSI, and H. J. RINGOLD, *J. org. Chem.* **26**, 871 (1961).

² R. E. MARKER and H. M. CROOKS Jr., *J. Amer. chem. Soc.* **64**, 1280 (1942).

³ S. BERNSTEIN, E. W. CANTRALL, and J. P. DUSZA, *J. org. Chem.* **26**, 269 (1961).

⁴ J. A. HOGG, F. H. LINCOLN, R. W. JACKSON, and W. J. SCHNEIDER, *J. Amer. chem. Soc.* **77**, 6401 (1955).

⁵ J. A. HOGG, G. B. SPERO, J. L. THOMPSON, B. J. MAGERLEIN, W. P. SCHNEIDER, D. H. PETERSON, O. K. SEBEK, H. C. MURRAY, J. C. BABCOCK, R. L. PEDERSON, and J. A. CAMPBELL, *Chem. and Ind.* **1958**, 1002.

⁶ A. BOWERS, *J. Amer. chem. Soc.* **81**, 4107 (1959).

Tab. I. Subcutaneous progestational activity – Clauberg assay^a

Compound	M.P. °C	$[\alpha]_D$ (CHCl ₃)	$\lambda_{\max}^{\text{MeOH}}$ m μ (e)	Approx. activity (Progesterone = 1)
16 α -Methylprogesterone (I)	135–138.5	+ 160°	240 (16900)	1
2 α , 16 α -Dimethylprogesterone (II) ^b	151–152	+ 169°	240 (14800)	inactive ^c
6 α , 16 α -Dimethylprogesterone (III)	113–116.5	+ 145°	242 (16100)	1
6 α -Fluoro-16 α -methylprogesterone (IV)	144–146	+ 18°	236 (14900)	1

^a Assays were performed by the Endocrine Laboratories, Madison (Wisconsin, U.S.A.) and the MacPhail modification of the Clauberg assay was used. ^b All new compounds recorded therein have been satisfactorily characterized by elemental analysis. ^c i.e. at a 1 mg total dose level.

Tab. II. Oral progestational activity – Clauberg assay^a

Compound	M.P. °C	$[\alpha]_D$ (CHCl ₃)	$\lambda_{\max}^{\text{MeOH}}$ m μ (e)	Approx. activity (17 α -Acetoxypregesterone = 1)
16 α -Methyl-17 α -hydroxypregesterone (V) ^b	180–183.5	+ 86°	242 (17200)	inactive ^c
16 α -Methyl-17 α -acetoxypregesterone (VI)	233–236	+ 80°	242 (15200)	1
16 α -Methyl-6-dehydro-17 α -acetoxypregesterone (VII)	212–214	+ 15°	283 (26100)	15
16 α -Methyl-1-dehydro-17 α -acetoxypregesterone (VIII)	227–228	+ 30°	244 (15000)	8
16 α -Methyl-1,6-bisdehydro-17 α -acetoxypregesterone (IX)	213–214	– 18°	221 (13900) 257 (12100) 299 (14000)	66

^a See Table I, footnote ^a. ^b See Table I, footnote ^b. ^c i.e. at a 1 mg total dose level.

one⁷ (V) was prepared from 16 α -methyl Reichstein's Substance S^{8,9} by reduction¹⁰ of the 21-tosylate of the latter. Compound V was also prepared in a seven-stage synthesis from 16 α -methylpregnenolone acetate² via the intermediate 5 α ,6 α -epoxy-17 α -hydroxy-16 α -methylpregnenolone benzoate [m.p. 228–230°, $[\alpha]_D = -49^\circ$ (CHCl₃)]. Acetylation of the 17 α -hydroxyl group provided the 17 α -acetate (VI). Bromination¹¹ of the enol ether of the latter with subsequent dehydrobromination with calcium carbonate in dimethylformamide¹¹ gave 16 α -methyl-6-dehydro-17 α -acetoxyprogesterone (VII). Dehydrogenation of VI and VII with 2,3-dichloro-5,6-dicyanobenzoquinone¹² gave 16 α -methyl-1-dehydro-17 α -acetoxyprogesterone (VIII)¹ and 16 α -methyl-1,6-bisdehydro-17 α -acetoxyprogesterone (IX), respectively.

Whereas a large increase in antiinflammatory activity has been associated with the introduction of a C16-methyl group in the corticoid series¹³, in no instance have we observed a pronounced increase in oral or subcutaneous progestational activity of the methylated progesterone over the corresponding desmethyl compound.

Zusammenfassung. Mehrere Derivate des 16 α -Methylprogesteron wurden synthetisiert und biologisch geprüft. Die Einführung einer C16 α -Methylgruppe ergab keine gesteigerte Progesteronaktivität in den getesteten Verbindungen.

Structural Basis of Nucleolar Function

Due to the lack of a definite concept as to whether structure or function is basic to the study of biological organization of cells^{1,2}, an integrated approach is the accepted operational choice of biologists to-day. Accordingly, studies on structure and function are expected to proceed concurrently in a balanced sequence. Recently, some valuable contributions have been made on the (hetero)chromatin-nucleolus-protein synthesis relationships³⁻⁷. Curiously, however, the exquisite details of the fine structure of nucleolus are still lacking.

In the present study⁸, some structural facets have been described by making a survey of extensive cytological changes that take place in the ophidian oocytes during their formation and development. The ophidian ovary was selected as being cytologically the least explored ovary. Moreover, the structural differentiation of the oocyte is perhaps the simplest. Yet, the biocytological processes cover a wide range of activities illustrative of cellular growth^{9,10}. The study has revealed some architectural patterns peculiar to nucleolar function.

Nucleolar behaviour. A recounting of the chain of events during the formation and maturation of the oocytes would reveal some interesting features of nucleolar organization. The early oocytes may be distinguished by the possession of a prominent coiled spireme within the nucleus during the prophase stages. At these stages, there is no indication of the presence of nucleolus. Later, a single nucleolus makes its appearance enmeshed in a loose reticulum of achromatic strands in close association with the suspended chromatin material (Figure 1). Thus, in the early stages of its formation, the nucleolus appears homogeneous. It is negative to the Feulgen reaction and has, therefore, no indications of DNA constituents. In advanced oocytes, commensurate with the growth of the oocyte, the nucleolus grows to a considerable size and soon begins to bud off several smaller bodies. The budding process of the nucleolus is shown in Figures 2 and 3. This process of actual budding has rarely been described. The nucleolar deriva-

S. BERNSTEIN, E.W. CANTRALL,
J. P. DUSZA, and J. P. JOSEPH

Organic Chemical Research Section, Lederle Laboratories, A Division of American Cyanamid Company, Pearl River (New York), June 6, 1961.

⁷ A compound assigned this structure was first described by K. HEUSLER, J. KEBRLE, C. MEYSTRE, H. UEBERWASSER, P. WIELAND, G. ANNER, and A. WETTSTEIN, *Helv. chim. Acta* **42**, 2043 (1959), with m.p. 217–220°, $\lambda_{\text{max}}^{\text{EtOH}}$ 242 m μ (ϵ 16600), $[\alpha]_D^{25} + 37.3^\circ$ (CHCl₃). Our characterization of this compound differs from that of the Swiss group but is in good agreement with that reported by the Syntex group¹.

⁸ J. A. EDWARDS, A. ZAFFARONI, H. J. RINGOLD, and C. DJERASSI, *Proc. chem. Soc.* **1959**, 87.

⁹ 16 α -Methyl-Reichstein's Substance S has also been prepared in this Laboratory from 16 α -methylpregnenolone acetate² via the intermediate methyl 3 β -hydroxy-16 α -methylpregnadien-5,17(20)-dien-21-oate.

¹⁰ A. BOWERS and H. J. RINGOLD, *J. Amer. chem. Soc.* **80**, 3091 (1958).

¹¹ L. H. KNOX, J. A. ZDERIC, J. PÉREZ RUELAS, C. DJERASSI, and H. J. RINGOLD, *J. Amer. chem. Soc.* **82**, 1230 (1960).

¹² D. BURN, D. N. KIRK, and V. PETROW, *Proc. chem. Soc.* **1960**, 14.

¹³ G. E. ARTH, D. B. R. JOHNSTON, J. FRIED, W. W. SPOONER, D. R. HOFF, and L. H. SARETT, *J. Amer. chem. Soc.* **80**, 3168 (1958). — E. P. OLIVETO, R. RAUSSER, A. L. NUSSBAUM, W. GEBERT, E. B. HERSHBERG, S. TOLKSDORF, M. EISLER, P. L. PERLMAN, and M. M. PECHET, *J. Amer. chem. Soc.* **80**, 4428 (1958).

tives are more or less of the same size. But exceptions to this may be found. The daughter nucleolar bodies come to lie near the nuclear membrane and are eventually extruded into the cytoplasm. In Figure 3, the much enlarged nucleolus is shown to be differentially stained. The central part is deeply stained while the outer part is lighter. This difference can be interpreted as due to variable concentrations of its constituent substances or to its duplex nature. Figures 2 and 3 also show in a clear manner the formation of daughter nucleolar bodies from the principal nucleolus, indicating that these daughter nucleolar bodies are not formed from any other source. The nucleolar organizers, which in earlier phases might be importantly concerned in the origination of nucleolus, do not seem visibly to participate in the formation of these nucleolar derivatives. There is no increase in the size of the nucleolar bodies after they reach the cytoplasm, neither do they multiply. They appear to be completely dissolved in the cytoplasm without having directly transformed into any kind of yolk material. The extruded nucleolar bodies provide the cytological mechanism for the exchange of certain chemical substances from the nucleus to the cytoplasm (Figure 4). Presumably, these activating chemical

¹ T. M. SONNEBORN, *Proc. Nat. Acad. Sci. U.S.A.* **46**, 149 (1960).

² L. PICKEN, *The Organization of Cells* (Oxford University Press, London 1960).

³ H. STICH, *Exper.* **12**, 7 (1956).

⁴ R. P. PERRY and M. ERRERA, *The Cell Nucleus* (Butterworths, London 1960), p. 24.

⁵ J. BRACHET, *Biochemical Cytology* (Academic Press, New York 1957).

⁶ T. CASPERSSON, *Cell Growth and Cell Function* (Norton, New York 1950).

⁷ W. S. VINCENT and E. BALTUS, *The Cell Nucleus* (Butterworths, London 1960), p. 18.

⁸ B. S. KAUSHIVA, Dissertation (University of Michigan, Microfilms 1949); *Biological Abstracts* **23**, 20256 (1949).

⁹ D. RUDNICK, *Cell Organism and Milieu*. 17th Symposium on Cells and their Environment. Society for the Study of Development and Growth (Ronald Press, New York 1959).

¹⁰ J. BRACHET, H. CHANTRENNE, and F. VANDERHAEGHE, *Biochim. biophys. Acta* **18**, 544 (1955).